

2022

Chem 250

Albert Aidoo

	MONDAY	TUES	WEDNESDAY	THUR	FRIDAY
	29 introduction	30 H atoms Inorg 1.2-1.3	31 aufbau, Slaters Rules Inorg 1.2-1.3	1	2 IE, EA, EN Inorg 1.6-1.7, 9.2
SEPT	5	6 <i>acac</i> <i>lab</i>	7 VSEPR, noble gas Inorg 2.1, 2.3, 18.1-18.2 <i>problems due</i>	8	9 VB Theory Inorg 2.4-2.6
	12 point groups Inorg 3.1, 3.3-3.4	13 <i>nickel</i> <i>lab</i>	14 point groups <i>Presentation 1</i>	15	16 <i>problems due</i> <i>TEST 1</i>
	19 ionic radii/bonding Inorg 3.10	20 <i>isomer</i> <i>lab</i>	21 Born Haber Inorg 3.11-3.14	22	23 ionic structures Inorg 4.1-4.3, 4.9 <i>acac lab due</i>
	26 Werner/Jorgenson Inorg 7.1-7.2, 7.7-7.10 <i>Literature Review 1 Due</i>	27 <i>lab</i>	28 crystal field Inorg 20.1 <i>problems due</i>	29	30 crystal field Inorg 20.1 <i>nickel lab due</i>
OCT	3 crystal field/coordination <i>Presentation 2</i>	4 <i>lab</i>	5 coordination Inorg 7.3-7.5 <i>problems due</i>	6	7 MO Theory Inorg 2.7-2.10 <i>isomer lab due</i>
	10 character tables Inorg 3.2, App. 4	11 <i>Pd</i> <i>lab</i>	12 ligand field Inorg 20.2 <i>problems due</i>	13	14 <i>TEST 2</i>
	17	18	19	20	21
	MIDTERM BREAK				
	24 EAN Inorg 22.1-22.4	25 <i>zeolite</i> <i>lab</i>	26 ligands Inorg 22.5-22.17	27	28 M-M bonds Inorg Box19.2, 22.20
	31 reactions Inorg 22.22-22.27	1 <i>lab</i>	2 Reactions <i>Presentation 3</i> <i>problems due</i>	3	4 kinetics Inorg 21.1-21.2a <i>palladium lab due</i>
NOV	7 metals/semiconductors Inorg 4.18-4.20	8 <i>LED</i> <i>lab</i>	9 <i>p-n</i> junctions Inorg 24.19-24.20	10	11 polyhedral, frameworks 14.10, 14.15, Box19.1, 24.11-12 <i>zeolite lab due</i>
	14 defects, oxides, magnetism 4.16-4.17, 20.8, 24.3-24.4	15 <i>project</i> <i>lab</i>	16 MOFs Inorg 24.11-24.12 <i>problems due</i>	17	18 <i>TEST</i> <i>LED lab due</i>
	21 plane groups, Escher SYM p. 1-5	22 <i>project</i> <i>lab</i>	23 3-D symmetry SYM p. 6-14 <i>problems due</i>	24	25 Thanksgiving
	28 International Tables SYM p. 15-17 <i>Literature Review 2 Due</i>	29 <i>project</i> <i>lab</i>	30 equiv position, ORTEP SYM p. 18-22 <i>problems due</i>	1	2 ORTEP SYM p. 22-23
DEC	5 Bioinorganic	6 Bioinorganic	7 Bioinorganic	8	9 <i>Presentation 4</i> On Green Chemistry
	12 Evaluation <i>Literature Review 3 Due</i>	13	14 no final (9 am)	15	16

Inorganic Chemistry by Weller, Overton, Rourke, Armstrong (7 ed).

SYM = *Symmetry and Crystallography* (2